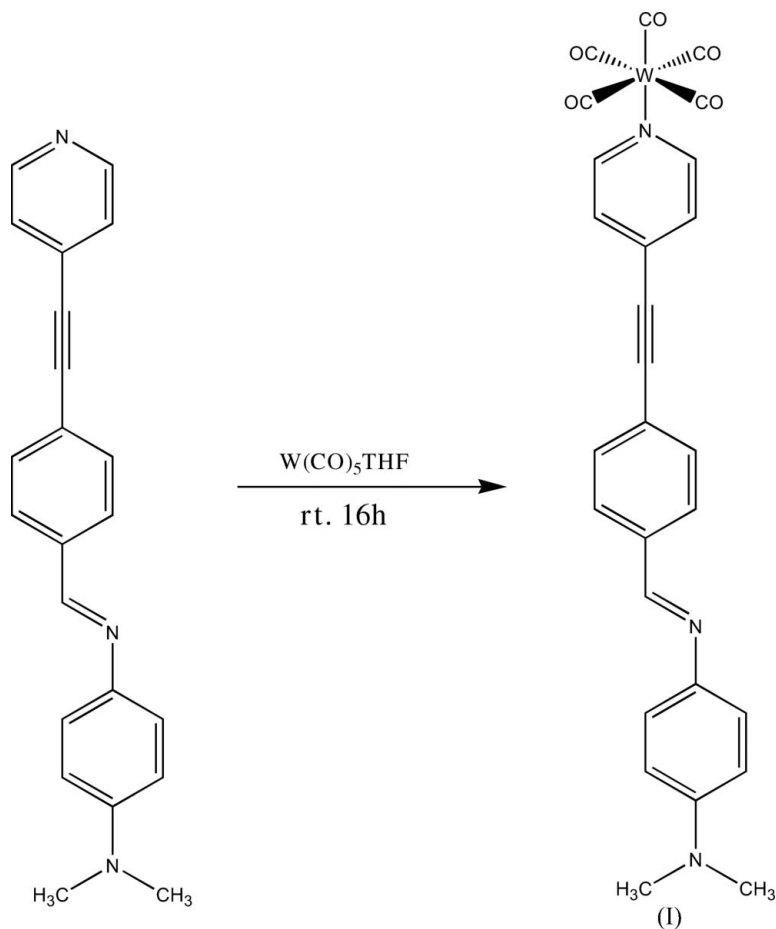


Ping-Hsin Huang^{a,b*} and
Yuh-Sheng Wen^a^aInstitute of Chemistry, Academia Sinica,
Nankang, Taipei, Taiwan 115, and ^bKang-Ning
Junior College of Medical Care and
Management, Taipei, Taiwan 114Correspondence e-mail:
pshuang@webmail.knjc.edu.tw

Key indicators

Single-crystal X-ray study
 $T = 100$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.022
 wR factor = 0.049
Data-to-parameter ratio = 13.9For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.Pentacarbonyl{*N,N*-dimethyl-*N'*-[4-(pyridin-4-yl)-
ethynyl- κN]benzylidene]benzene-1,4-diamine}-
tungsten(0)The asymmetric unit of the title compound, $[\text{W}(\text{C}_{22}\text{H}_{19}\text{N}_3)(\text{CO})_5]$, contains two molecules. The geometry at the W atom is approximately octahedral, with the *cis* bond angles in the range $86.11(13)$ – $92.75(11)^\circ$. The axial carbonyl bonds [$\text{W}-\text{C} = 1.960(4)$ and $1.967(4)$ Å] are shorter than the equatorial carbonyl bonds, which lie in the range $2.014(4)$ – $2.047(4)$ Å.Received 8 January 2007
Accepted 16 January 2007

Comment

There is widespread interest in π -conjugated molecular frameworks because these materials may be used as molecular wires (Woitellier, 1989) and opto-electronic devices. The title compound, (I), containing a tungsten carbonyl unit as an electron acceptor with an end-capping organic electron donor and benzene ring in the conjugation chain, has great potential as an opto-electronic material (McCullough, 1998).

distorted octahedral environment, bound to five carbonyl groups and one *N,N*-dimethyl-*N'*-[4-(pyridin-4-ylethynyl- κ *N*)benzylidene]benzene-1,4-diamine ligand.

The bond distances and angles involving the W atoms are listed in Table 1. The W–C_{axial} bonds are shorter [W–C = 1.960 (4) and 1.967 (4) Å] than the equatorial W–CO bonds which are in the range 2.014 (4)–2.042 (3) [angles 86.11 (13)–92.75 (11)°]. The dihedral angle between the C19–C24 and N11/C12–C16 rings is 23.58 (1)° and that between the N11/C12–C16 and C26–C31 rings is 2.21 (1)° [10.03 (1) and 36.91 (1)° for the other molecule]. The two benzene rings form a dihedral angle of 21.65 (1)° [46.80 (1)° for the other molecule]. No significant hydrogen-bonding interactions are observed in the crystal structure.

Experimental

A tetrahydrofuran (THF) solution (100 ml) of W(CO)₅(THF) prepared from W(CO)₆ (100 mg, 0.28 mmol) was transferred to a flask containing *N,N*-dimethyl-*N'*-[4-(pyridin-4-ylethynyl- κ *N*)benzylidene]benzene-1,4-diamine (78 mg, 0.24 mmol). The solution was stirred at room temperature for 16 h, and the solvent was removed under vacuum. The residue was chromatographed through silica gel using a THF/hexane (1:10) mixture as eluant. The compound was obtained as a red solid in 48% yield. FAB–MS: *m/e* 650 (*M* + 1)⁺. Analysis calculated for C₂₇H₁₉N₃O₅W: C 49.94, H 2.95, N 6.47%; found: C 49.50, H 2.88, N 6.38%.

Crystal data

[W(C ₂₂ H ₁₉ N ₃)(CO) ₅]	<i>V</i> = 2564.09 (7) Å ³
<i>M_r</i> = 649.30	<i>Z</i> = 4
Triclinic, <i>P</i> $\bar{1}$	<i>D_x</i> = 1.682 Mg m ⁻³
<i>a</i> = 12.2258 (2) Å	Mo <i>K</i> α radiation
<i>b</i> = 12.5046 (2) Å	<i>μ</i> = 4.55 mm ⁻¹
<i>c</i> = 17.3157 (3) Å	<i>T</i> = 100.0 (1) K
<i>α</i> = 102.595 (1)°	Plate, red
<i>β</i> = 95.362 (1)°	0.3 × 0.25 × 0.08 mm
<i>γ</i> = 93.249 (1)°	

Data collection

Bruker SMART CCD area-detector diffractometer	39209 measured reflections
<i>φ</i> and <i>ω</i> scans	9009 independent reflections
Absorption correction: multi-scan (SADABS; Bruker, 2001)	7221 reflections with <i>I</i> > 2σ(<i>I</i>)
<i>T_{min}</i> = 0.493, <i>T_{max}</i> = 0.986 (expected range = 0.348–0.695)	<i>R_{int}</i> = 0.060
	<i>θ_{max}</i> = 25.0°

Refinement

Refinement on <i>F</i> ²	<i>w</i> = 1/[σ ² (<i>F_o</i> ²) + (0.0204 <i>P</i>) ²]
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.022	where <i>P</i> = (<i>F_o</i> ² + 2 <i>F_c</i> ²)/3
<i>wR</i> (<i>F</i> ²) = 0.049	(Δ/σ) _{max} = 0.002
<i>S</i> = 0.99	Δρ _{max} = 0.54 e Å ⁻³
9009 reflections	Δρ _{min} = -0.94 e Å ⁻³
650 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: 0.00043 (5)

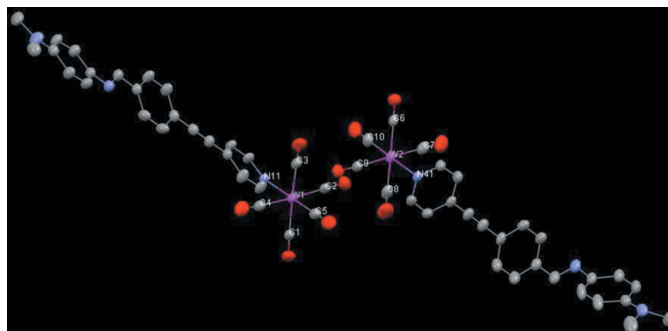


Figure 1 The asymmetric unit of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme employed.

Table 1

Selected geometric parameters (Å, °).

W1–C5	1.960 (4)	W2–C10	1.967 (4)
W1–C3	2.014 (4)	W2–C8	2.028 (4)
W1–C4	2.032 (4)	W2–C7	2.035 (4)
W1–C2	2.037 (4)	W2–C9	2.037 (4)
W1–C1	2.042 (3)	W2–C6	2.047 (4)
W1–N11	2.273 (2)	W2–N41	2.267 (2)
C5–W1–C3	86.11 (13)	C4–W1–C1	91.34 (13)
C5–W1–C4	89.60 (14)	C2–W1–C1	91.86 (12)
C3–W1–C4	87.78 (14)	C5–W1–N11	177.46 (11)
C5–W1–C2	89.67 (14)	C3–W1–N11	92.18 (11)
C3–W1–C2	88.97 (13)	C4–W1–N11	88.46 (11)
C4–W1–C2	176.70 (12)	C2–W1–N11	92.18 (11)
C5–W1–C1	88.92 (13)	C1–W1–N11	92.75 (11)
C3–W1–C1	174.96 (12)		

H atoms were positioned geometrically and treated as riding atoms, with C–H = 0.93–0.96 Å and *U_{iso}*(H) = 1.2*U_{eq}*(C).

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

This work is partially supported by the Institute of Chemistry, Academia Sinica, and Kang-Ning Junior College of Medical Care and Management.

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